



RESEARCH ARTICLE

MICROWAVE-ASSISTED SYNTHESIS OF GOLD NANOPARTICLES USING NEEM LEAVES
(AZADIRACHTA INDICA)

*¹Abha Anand and ²Murthy Chavali

¹Department of Biotechnology, Jawaharlal Nehru Technological University Hyderabad
(JNTUH) 500 085 Telangana State, India

²MCETRC, Chinnaravuru, Tenali, Guntur 522 201 AP, India

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ABSTRACT

Ecological generation and isolation of heavy metals such as Ag, Au, Cu, etc by Phytoextraction one of the best methods in biotechnology and nanotechnology field for various applications for human use. Since Gold (Au) is one of the expensive heavy metal, so Gold Nanoparticles (GNPs) or compounds are also available in market at a very high price. Therefore the isolation of GNPs by Phytoextraction methods cost comparative very low. GNPs are easily prepared using neem (Azadirachta indica) leaves by phytoextraction procedure using HAuCl₄. This research work provides a more possible option for generation of GNPs in lesser time, very low cost and non-toxic procedure without using large and complex apparatus. This paper deals with the generation of eco-friendly and cost effective gold nanoparticles by using leaf extract of neem (Azadirachta indica) as reducing agent in short period of time by microwave assistance showing improvement particularly for reduction of time, modulation of size and shape.

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INTRODUCTION

Phytoextraction is a secondary procedure of phytoremediation in which plants remove wanted and unwanted elements such as metal, heavy metal, compound and other contaminants from soil or water. Phytoremediation is an environmentally forthcoming, protected and inexpensive technique used to get rid of pollutants from an environment (Zhang *et al.*, 2010). Generally, heavy metals, metals which are extracted have a high density, may be toxic to organisms even at relatively low concentrations and are toxic to the plants as well. This is one of the effective methods in the field of biotechnology and nanotechnology for ecological generation of heavy metals nanoparticles such as gold, Hg, Cu, etc for various applications for human use at very low cost. It is a cost effective, long-term, environmentally and aesthetically friendly method of immobilizing/stabilizing and transferring contaminants such as metals, pesticides and chlorinated hydrocarbon without causing any disturbance to the land (Zhang *et al.*, 2010; Tai *et al.*, 2011). Au NPs have been synthesized using various fruit and plant extracts (Tai *et al.*, 2011).

Azadirachta indica is commonly known as Neem belonging to family Meliaceae. It is typically grown in tropical, semi-tropical regions and in islands located in the southern part of Iran. Neem oil is obtained from its fruits and kernel of seeds (Paul, 2011). Neem is one of a very few shade-giving trees that thrive in drought-prone areas. The trees are not at all insubstantial about water quality and thrive on the merest trickle of water, whatever the quality (Nahak, 2011). Neem products are believed as a major component by Siddha and Ayurvedic (Lekshmi *et al.*, 2012) practitioners to be Anthelmintic, antifungal (Anjali *et al.*, 2013; Shrivastava, 2014), antidiabetic (Patil *et al.*, 2013; Dholi *et al.*, 2011), antibacterial (Yerima *et al.*, 2012), antiviral (Arumugam *et al.*, 2014), contraceptive, and sedative (Baligar *et al.*, 2014). Neem oil is also used for healthy hair, to improve liver function, detoxify the blood (Nahak, 2011), and balance blood sugar levels (Nahak, 2011; Patil *et al.*, 2013; Dholi *et al.*, 2011). The biopesticide produced by extraction from the tree seeds contains Limonoids. Plant chemical ingredients concentration is limited by the soil concentration and by suitable ligands. This is a natural process in which environmentally occurring chemicals will cause plants to accumulate Au. This is also a known process biogeochemical exploration (Kiranmai *et al.*, 2011).

*Corresponding author: Abha Anand,
Department of Biotechnology, Jawaharlal Nehru Technological University
Hyderabad (JNTUH) 500 085 Telangana State, India.

Gold nanoparticles (GNPs) are used for various application viz, gold-colloid paints, electronics, as an industrial catalyst and biomedical purposes. Development of GNPs by this phytoextraction method is more cost-effective gold recovery system based on nanotechnology (Tai *et al.*, 2011). Recently many researchers are providing much more prominence on biological or plant-mediated synthesis of nanoparticles as it is simple, eco-friendly and also low cost. Despite the fact that it has previously been reported the plant-mediated synthesis or phytoextraction of gold nanoparticles such as Terminalia catappa (Ankamwar, 2008), tea (Satish *et al.*, 2009), Cymbopogon citratus, lemongrass tree, (Shiv Shankar *et al.*, 2004) Allium cepa (Parida *et al.*, 2011) and fruit extracts (Tai *et al.*, 2011) but still upcoming researches are there.

Research on neem (*Azadirachta indica*) (Birendra Kumar Bindhani, 2014) shows synthesis is time consuming almost 3-4 days for isolation of gold nanoparticles, therefore this research work shows improvement particularly for reduction of time for synthesis of nanoparticles by microwave assistance, modulation of size and shape. This experiment is a very appealing study to know the nature of nanoparticles which is produced by using leaves extracts. The selection of neem (*Azadirachta indica*) plant extracts for this study because of abundant availability and its numerous biomedical applications and so on (Birendra Kumar Bindhani, 2014). The main active ingredients are chemically similar and biogenetically derivable from a tetracyclic terpenes. These are also called limonoids (azadirachtin, melianthrol, salanin etc.) Neem limonoids belong to nine basic structures: azadiron (from seed oil), amoorastatin (from fresh leaves), gedunins (from kernels), salannin (from fresh leaves and seeds), and azadirachtin (from seed oil) (Paul, 2011). The activated neem leaves consist of mainly three different kinds of phenolic compounds such as 4-chlorophenol, 4-nitrophenol and phenol (P) was studied from fixed bed mode (Birendra Kumar Bindhani, 2014). 4-nitrophenol is a white to light yellow crystalline solid. Contact may severely irritate the skin and eyes. Poisonous by ingestion and moderately toxic by skin contact (Manohar *et al.*, 2016).

Objective: The objective of the proposed research work is to develop a simple, green method for the isolation and separation of Gold (Au) nanoparticles (NPs) from the leaves extract of neem (*Azadirachta indica*). Unlike other biological methods for nanoparticles synthesis, the uniqueness of this method lies in its fast synthesis rates (~3 min for AuNPs) once the extract is ready through phytoextraction of neem and the capability to tune the nanoparticle size (and subsequently their catalytic activity (Manohar *et al.*, 2016)) via the extract concentration used in the experiment. The phenolic ions (p-nitrophenol) present in the extract are largely responsible for the rapid reduction rates of Au³⁺ ions to AuNPs. Efficient reduction of 4-nitrophenol (4-NP) to 4-aminophenol (4-AP) in the presence of AuNPs was observed and found to depend upon the nanoparticle size or the leaves extract concentration used for isolation of gold nanoparticles. This work aimed to the synthesis of the gold nanoparticle using neem leaves (*Azadirachta indica*) within an anticipated time period i.e. within a day and finally, gold nanoparticles were characterized through UV-Visible and SEM techniques.

MATERIALS AND METHODS

Plant material: The fresh leaves of Neem (*Azadirachta indica*) were collected in the month of September from

Balanagar, Hyderabad. Leaves were rinsed and repeatedly washed with water properly to remove dust particles followed by air dried at room temperature.

Chemical and Reagents: All chemicals used were analytical reagent grade and the solutions were prepared with Millipore purified distilled water. H₂SO₄·XH₂O (Tetrachloroauric acid), ethanol (96%) was obtained from Sigma Aldrich was used to carry out experimental activities. All glassware such as a conical flask, measuring cylinder, test tubes burette, beaker, Petri dishes were thoroughly washed and rinsed with purified water followed by drying properly for the experiment. The experiments are carried out by using re-usable flat bottom glass cuvettes for absorbance mode.

Preparation of Ethanolic Extract: Neem (*Azadirachta indica*) leaves (150 g) were separated, roughly chopped into pieces and again dried up by keeping in drying oven at temperature 60-70°C. Dried Leaves were blended for extraction by Soxhlet extraction method. 96% ethanol was used as a solvent for extraction. The analytical balance was used for weighing neem leaves and heater Mantle was used for heating the solution during Soxhlet extraction time. Soxhlet apparatus was used for the extraction process were maintained with temperature 70-90°C till the extract is colour less. Filter papers were used for filtration of extract followed evaporation to the concentrated filtrate. 100 ml conical flasks and 100 ml volumetric flasks, beakers and pipette were used for measuring and solution preparation.

Phytoextraction of Gold Nanoparticles: Washed neem leaves (20g) were chopped finely for preparation of the stock which was used for reduction reaction (from Au³⁺ ions to Au⁰). Chopped neem leaves were kept in a 500 ml conical flask with 60 ml of Millipore distilled water. The solution was boiled for 20min. Based on literature review and suggestion of expert in department of Pharmacognosy, this research was planned as 100ml of 10⁻³ M aqueous chloroauric acid (HAuCl₄) was mixed with 0.3 ml of stock solution and kept inside domestic microwave (LG iwave working at 100% power of 600 watt (W)) for irradiation for 90 Sec. as explained in fig.1. The colour of the solution was changed to a red colour within a few minutes (2-3 minutes). The reaction times were determined by the change in colour of the solution. The reaction time 2-3 min was adequate for the complete conversion of colloidal Gold Nanoparticles. This stabilization of nitrophenol is not possible in the case of *m*-nitrophenol because there is no chance via resonance for delocalization onto the nitro group. Resonance is due to delocalization of negative charge at Ortho and Para position only whereas there is no delocalization at Meta position

RESULTS AND DISCUSSION

UV-Vis Spectrometric Method: Labindia UV3000 Spectrophotometer was used to take absorption of aqueous solution in the UV-visible spectral region. This instrument has holographic grating greatly reducing stray light of the instrument and making the analysis more accurate. The absorption during the observable range directly affects the perceived colour of the concerned chemicals. In this region of the electromagnetic spectrum, molecules undergo electronic transitions (Birendra Kumar Bindhani, 2014). Here in the experiment, 5 different concentrations of HAuCl₄ with neem extract and blank as shown in Fig. 2

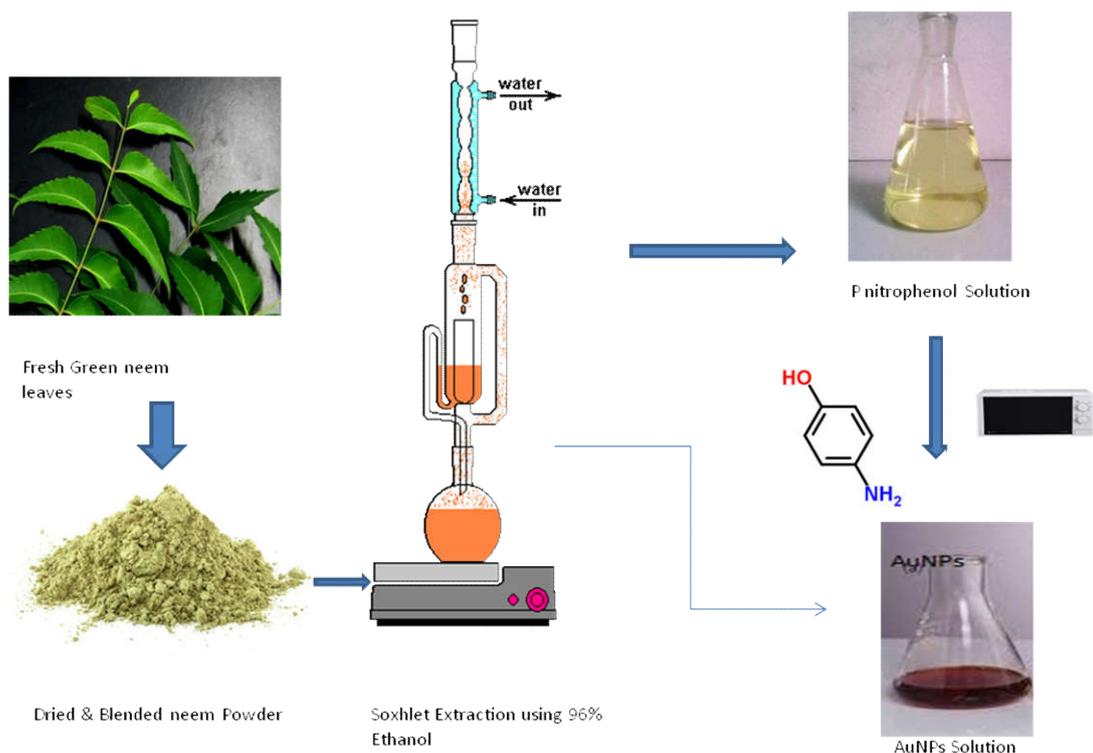


Fig. 1. Schematic diagram of Phytoextraction of Gold nanoparticles using HAuCl_4 ; Soxhlet extraction procedure was assembled using 96% ethanol for obtaining a colourless solution of p nitrophenol. Conversion of colour from pale yellow to red with the addition of HAuCl_4 is due to delocalization of negative charge of nitrophenol

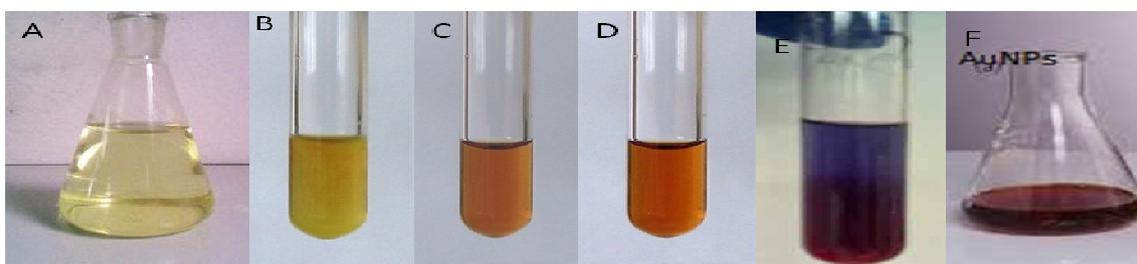


Fig.2. A Neem (*Azadirachta indica*) leaves extract, B-Gold (HAuCl_4) Aqueous Solution[Au], C- Au+1ml extract, D- Au+3ml extract, E-Au+5ml extract, F-Au+10ml extract of neem

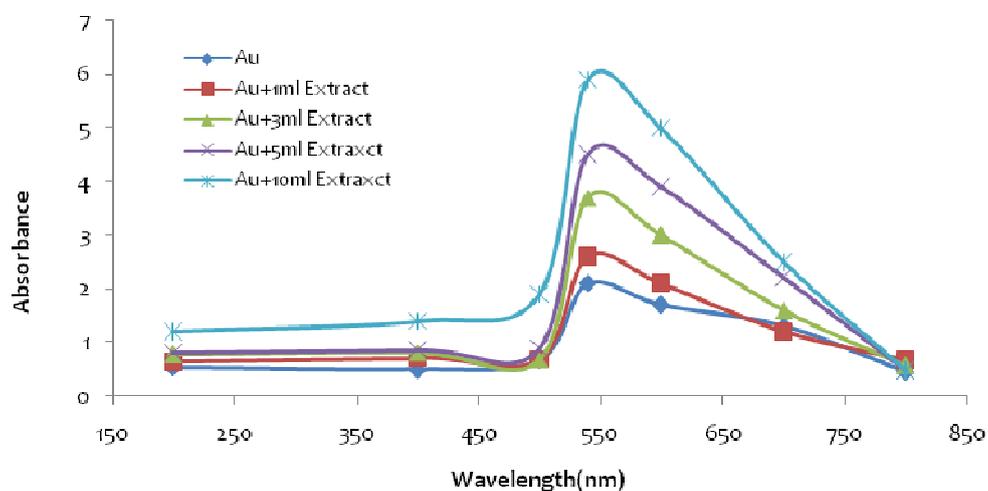


Fig.3. UV-Vis spectra of gold nanoparticles aqueous solution synthesized by adding a different volume of Neem (*Azadirachta indica*) extract to a fixed volume (10 ml) of HAuCl_4 solution (10–3 M)

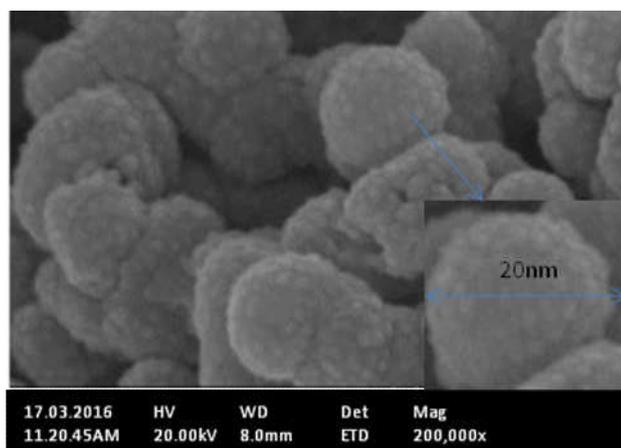


Fig.4. SEM images of the gold nanoparticles colloids (Spherical in shape & Size 10-20nm)

Absorbance's were observed at a different wavelength as shown in fig.3. The UV-Vis near infrared absorption spectra was used in aqueous suspension for the synthesis of gold nanoparticles and optical properties were exhibited closely associated with their shapes of such aqueous suspension metal nanoparticles (Birendra Kumar Bindhani, 2014). The peak (transverse plasmon resonance band) was observed at 540 nm shifted towards a higher wavelength, confirming a redshift along with amplified absorbance intensity. The UV-vis spectrum (Figure 3A) also showed a very broad band towards the longer wavelength (IR) region.

The transform of colour was observed after the completion of each reaction of 5 samples under microwave irradiation is shown in Fig. 3A-F. The surface plasmon resonance (SPR) bands were observed using UV-Vis spectroscopy for the isolated colloidal suspensions of AuNPs clearly shows the gradual decrease in the intensity of absorption of SPR band. As the moving from the SPR band corresponding to a sample of Au NPs, the intensity of absorption decreases proportionally to wavelength revealing the decrease in particles size of the sample (Manohar, 2016). The reaction was observed as the conversion nitrophenol to p-aminophenol may be attributed due to disturbance of morphology of nanoparticles and the agglomeration of nanoparticles. The gradual reduction of p-nitrophenol to p-aminophenol was monitored at regular time intervals by taking 0.5 mL of the reaction mixture and diluting it up to 50 mL with distilled water and recording its UV-Visible spectrum at ambient condition.

SEM (Scanning Electron Microscopy) Measurements:

Scanning electron microscopy measurement was carried out by using Hitachi S - 4700 SEM machine. Samples were prepared on a carbon-coated copper grid as a thin and small oval-shaped film. Removal of the extra solution was done using an absorbent paper before keeping the sample under a mercury lamp for 3-5 min for drying. The colloidal gold nanoparticles obtained from the reaction of HAuCl_4 with extract of neem (*Azadirachta indica*) followed by microwave irradiation are primarily observed morphologically spherical in shape with size 10-20nm as shown in fig.4 below.

Conclusion

In summary, this experimental work shows a simplistic, fast and microwave-assisted process for the synthesis of gold nanoparticles from neem leaves (*Azadirachta indica*).

The colloidal AuNPs were obtained within 3 minutes under microwave irradiation and the particle sizes were observed in the range of 10–20 nm. Furthermore, the reaction is a reduction of p-nitrophenol to p-aminophenol and finally resultant AuNPs using HAuCl_4 . It can be concluded that the Au NPs synthesized from the HAuCl_4 show that microwave assisted synthesis is a faster reduction of p-nitro phenol to p-aminophenol than the other activities, with proper modulation of size and shape of nanoparticles.

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