



ISSN: 0975-833X

## RESEARCH ARTICLE

### SYNTHESIS, CHARACTERIZATION AND ANTIBACTERIAL ACTIVITY OF 2-(1, 2, 3 -BENZOTRIAZOL-1-YL-ACETATE) THIAZOLIDINE ACETIC ACID ETHYL ESTER USING DIETHYL ACETYLENE DICARBOXYLATE (DEAD) WITH THIOSEMICARBAZONE DERIVATIVES

<sup>1</sup>Nandhikumar, R., <sup>2\*</sup>Dr. Subramani, K. and <sup>3</sup>Dr. Syed Shafi, S.

<sup>1</sup>Bharathiyar University, Coimbatore, Tamilnadu, India

<sup>2</sup>Department of Chemistry, Islamiah College, Vaniyambadi, Tamilnadu, India

<sup>3</sup>Thiruvalluvar University, Vellore, Tamilnadu, India

#### ARTICLE INFO

##### Article History:

Received 15<sup>th</sup> December, 2013

Received in revised form

30<sup>th</sup> January, 2014

Accepted 04<sup>th</sup> February, 2014

Published online 25<sup>th</sup> March, 2014

##### Key words:

Benzotriazole, Condensation,  
Thiosemicarbazide, Thiazolidine,  
Anti-bacterial activity.

#### ABSTRACT

In the present investigation newer and simple synthetic methods of 2-(1, 2, 3- benzotriazol-1-yl-methyl) thiazolidine acetic ethylester is described. Benzotriazole 1 is converted to carbothioamide 3 by reaction with ethylchloroacetate followed by thiosemicarbazide. The compound 3 is converted to corresponding thiazolidine compound by treatment with Diethyl Acetylene Dicarboxylate (DEAD) With Thiosemicarbazone Derivatives. Structural elucidation is accomplished by IR, and <sup>1</sup>H NMR spectral data of the synthesized compounds. Based on the antibacterial studies of the compound 1 and 2, it can be concluded that compound 2 showed (plate-1) high activity against *Escherichia coli* (gram-negative bacteria) at 100 µg concentration then compound 1.

Copyright © 2014 Dr. Ashfaque Hassan and Dr. Shifan Khandey. This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

#### INTRODUCTION

Azoles are most important in the history of heterocyclic chemistry and also extensively as important synthons in organic synthesis. The versatile chemotherapeutic activities of azoles, a significant amount of research activity have been directed towards this class. Synthesis and activity of benzotriazole derivatives as antiprotozoal agents (Kopanska *et al.*, 2004) (inhibitors of *Acanthamoeba castellanii*) have been reported in the literature. Benzotriazole acts as raw materials in many organic syntheses (Purohit and Srivastava 1992; Krasavin 2005) and has proven to be fertile source of Pharmaceutical agents such as antimicrobial (Al-omran *et al.*, 2002), anticonvulsant, anti-inflammatory (Dawood *et al.*, 2006), antitumor (Al-Soud *et al.*, 2003) etc. Some derivatives of benzotriazoles are reported as agonists of peroxisome proliferator activated receptors (Sparatore *et al.*, 2006). Synthesis and biological activity of 1H-benzotriazol analogues as inhibitors of the NT pasc / helicase and of some related Flavivirade has been extensively investigate (Bretner *et al.*, 2005). Thiazolidinones have a broad spectrum of pharmacological properties i.e. antibacterial (Altintas *et al.*, 2005), antifungal (Liu *et al.*, 2000), anti-inflammatory (Vazzana *et al.*, 2004), anticonvulsant (Gursoy and Terzioglu 2005) etc. Thiazolidinones, substituted at the position two, its

derivatives and analogues exhibit unusually high in vitro activity against *Mycobacterium tuberculosis* (Sobin 1952). Several derivatives of alkoxyphthalimide have been synthesized (Sharma *et al.*, 2006; Banu *et al.*, 2000) and reported to demonstrate a wide range of pharmacological activities i.e. anticancer, antimalarial (Ure and Perassalo 2000), antiepileptic (Singh *et al.*, 2004) etc.

#### RESULTS AND DISCUSSION

The key intermediate used for the synthesis of both series of the final compounds was 2-(1,2,3- benzotriazol-1-yl-acetate)hydrazine carbothioamide 3, which in turn was prepared by the reaction of 1H-benzotriazole 1 with ethylchloroacetate in the presence of K<sub>2</sub>CO<sub>3</sub> as a base, followed by condensation with thiosemicarbazide. Formation of 3 was confirmed by the presence of N-H stretching peaks at 3378 and 3237 cm<sup>-1</sup> in IR and a multiplet at 8.3 for NH.NH.C=S.NH<sub>2</sub> group in <sup>1</sup>H NMR spectra. Then the compound reacted with DEAD to form heterocyclisation. And The synthesised compounds were tested for antibacterial activity against *Escherichia Coli* (gram-negative bacteria) and *Staphylococcus aureus* (gram- positive bacteria).

#### Experimental section

Melting points were taken in open capillary tubes and are uncorrected. Infrared spectra were recorded on Nicolet 380 -

\*Corresponding author: Dr. Subramani, K.

Department of Chemistry, Islamiah College, Vaniyambadi, Tamilnadu, India

FT-IR Spectrophotometer using KBr Pellets ( $\gamma$  max in  $\text{cm}^{-1}$ ). Thin layer chromatography (TLC) was performed using glass plates, coated with Silica gel (ACME) of 0.25mm thickness. Spots were visualized using iodine chamber. Usual workup and column chromatographic purification. (60-120 mess Ethyl acetate: petroleum ether). The solvents and reagents used for the synthesis were purified by the standard methods. Petroleum ether refers to the fraction of b.p 60-80. Anhydrous sodium sulphate was used as the drying agent. The synthesized compounds were tested for antibacterial activity.

### Synthesis of Benzotriazole

Dissolved 10.8g of (0.1 mol) of 0-phenylenediamine (1) in a mixture of 12g (11.5ml 0.2mol) of glacial acetic acid 30ml of water contained in a 250ml beaker and then added 7.5g (0.11mol) of sodium nitrite in 15ml of water in one portion, continue stirring for 15 minutes. Collect by vacuum filtration the pale brown solid and dissolve boiling water with charcoal and filter, dried and Usual workup and column chromatographic purification (Ethylacetate : Petroleum ether). m.p 95 - 99 c. Swati Ojha (2007) reported the same melting point. Synthesis of Ethyl - (1,2,3 benzotriazol) - 1 - yl acetate (I) To a solution of benzotriazole (1) (0.01mole, 1.19g) in acetone, ethylchloroacetate (0.01mole, 1.06ml) was added drop-wise and  $\text{K}_2\text{CO}_3$  (0.01mole, 2.76g) was used as a base. The reacting mixture was refluxed for 7 hr. On a water bath and filtered hot. Solvent was evaporated from the filtrate to yield the product as white, shining crystals. Recrystallization was carried out from ethanol. The purity was checked on a TLC (Silica gel). Using a mixture of benzene and methanol in the ratio 3:1 by volume Usual workup and column chromatographic purification (Ethylacetate: Petroleum ether) Yield 66%; m.p. 72 C IR (KBr,  $\text{cm}^{-1}$ ): 2987 (C-H str., CH<sub>3</sub>), 2934 (C-H str., CH<sub>2</sub>), 1742 (C=O str.), 1600 (C=N str.), 1028.82 (C-O str.); <sup>1</sup>H NMR (CDCl<sub>3</sub>,  $\delta$ ): 7.65 (m, 4H, Ar-H), 4.20 (q, 2H, COOCH<sub>2</sub>CH<sub>3</sub>), 3.65 (s, 2H, NCH<sub>2</sub>), 1.28 (t, 3H, COOCH<sub>2</sub>CH<sub>3</sub>)

### Synthesis of 2 - (1,2,3 - benzotriazol - 1 yl - acetate- hydrazine carbothioamide (II)

An equimolar mixture of (I) (0.01mole, 2.5g) and thiosemicarbazide (0.01mole, 0.9g) in acetone was refluxed for 8-10 hr. The reaction mixture was allowed to cool and the obtained yellow solid was recrystallised from alcohol. The purity of the compound was checked on a TLC (Silica gel) using a mixture of benzene and methanol 3:1 by volume. The spots were identified with iodine vapour in an iodine chamber Usual workup and column chromatographic purification (Ethylacetate: Petroleum ether) (Yield 81%); m.p

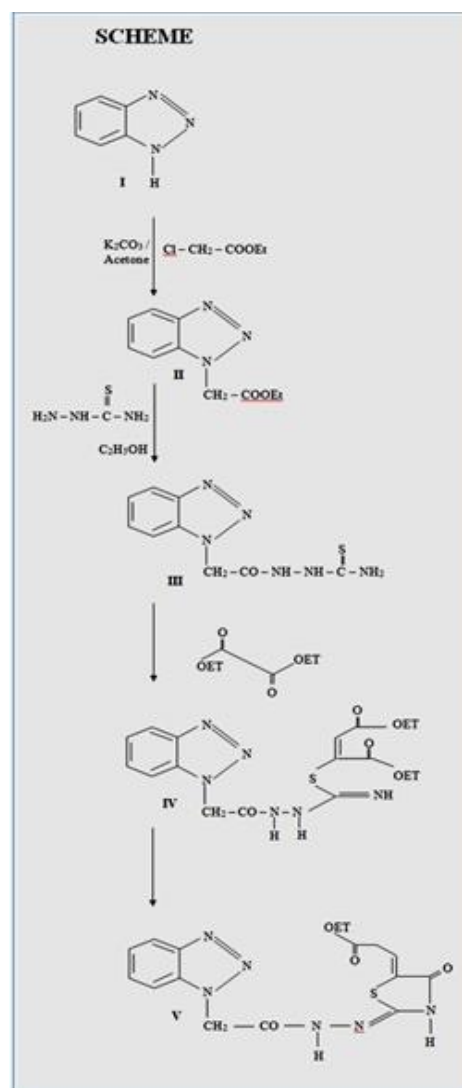
### Synthesis of 2 - (1,2,3 - benzotriazol - 1 -yl - acetate) thiazolidine acetic acid ethyl ester(III)

To a solution of 2-(1,2,3-benzotriazol-1-yl)-acetate)-ydrazine carbothioamide (2.50g) 0.01mol and diethyl acetylene dicarboxylate (DEAD) (0.01mol, 0.3ml) and

ethyl acetate (100ml) was added to a solution. The solution was stirred at ambient temperature for 3h. The resulting yellow precipitate was filtered, washed with ethyl acetate, a yellow solid was separated which was than recrystallized from ethanol water. The purity of a compound was checked on a TLC (Silica gel) using a mixture of benzene and methanol 3:1 by volume. The spots were identified with iodine vapour in an iodine chamber. Usual workup and column chromatographic purification (Ethylacetate : Petroleum ether) (Yield 95%) m.p. 242 C. IR (KBr,  $\text{cm}^{-1}$ ): 3449, 3380 (N-H str.), 1787.20, 1662.67 (C=O str.), 1603.71 (C=N str.), 695 (C-S-C str.); <sup>1</sup>H NMR (CDCl<sub>3</sub>,  $\delta$ ): 8.13 (s, 1H, CONH), 7.1-7.5 (m, 4H, Ar-H), 5.8 (s, 1H, NH of thiazolidinone ring), 4.1 (s, 2H, CH<sub>2</sub>), 3.8 (s, 2H, NCH<sub>2</sub>).

### Bacterial activity

Based on the antibacterial studies of the compounds 1 and 2, it can be concluded that compound 2 showed (Plate - 1) high activity against Escherichia coli (gram-negative bacteria) at 100  $\mu\text{g}$  concentration then compound 1.



### Acknowledgement

The authors are thankful to the Dr. S. Syed Shafi Department chemistry, Thiruvalluvar University, Vellore (Tamilnadu) And

Islamiah college Vaniyambadi (Tamilnadu) for providing laboratory facilities and providing spectral and analytical data.

## REFERENCES

- Al-omran F, Mohareb R M and El-Khair A A, J Heterocyclic Chem, 39(5), 2002, 877.
- Al-Soud Y A, Al-Masoudi N A and Ferwanah Ael-R, Bioorg Med Chem, 11(8), 2003, 1701.
- Altintas H, Ates O, Birtoksoz S, OtukG, Uzun M and Satana D, Turk J Chem, 29, 2005, 425.
- Banu T, Rajora S, Khatri D and Talesara G L, J Indian Chem Soc, 77, 2000, 300.
- Bretner M, Baier A, Kopanska K, Najda A, Schoof A, Reinholz M, Lipniacki A, Piasek A, Kulikowski T and Borowski P, Antivir Chem Chemother, 16, 2005, 315.
- Dawood K M, Abdel-Gawad H, Rageb E A, Ellithey M and Mohammed H A, Bioorg Med Chem, 14(11), 2006, 3672.
- Demschroder R C and Peterson W D, Org Syn Coll Vol, III, 1955, 106.
- Gursoy A and Terzioglu N, Turk J Chem, 29, 2005, 247.
- Kopanska K, Najda A, Zebrowska J, Chomicz L, Piekarczyk J, Myjak P and Bretner M, Bioorg Med Chem, 12, 2004, 2617.
- Krasavin M, Pershin D G, Larkin D and Kravchenko D, Synthetic Communication, 35, 2005, 2587.
- Liu H L, Li Z and Anthonsen T, Molecules, 5, 2000, 1055.
- Purohit M and Srivastava S K, Indian J Pharmaceutical Sciences, 54(1), 1992, 25p
- Sharma R, Nagda D P and Talesara G L, ARKIVOC, i, 2006, 1.
- Singh B, Mehta D, Baregama L K and Talesara G L, Indian J Chem, 43B, 2004, 1306.
- Sobin B A, J Am Chem Soc, 74, 1952, 2497.
- Sparatore A, Godia C, Perrino E, Romeo S, Stales B, Fruchart J C and Crestani M, Chemistry and Biodiversity, 3, 2006, 385.
- Ure J A and Perassalo M, J Neuro Sc, 1, 2000, 177.
- Vazzana I, Terranova E, Mattioli F S and Sparatore F, ARKIVOC, 5, 2004, 364.

\*\*\*\*\*