



RESEARCH ARTICLE

ZINC DOPED CERIUM OXIDE THIN FILMS PREPARED BY  
SPRAY PYROLYSIS METHOD

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ABSTRACT

Zinc doped Cerium Oxide thin films were prepared by spray pyrolysis technique. The CeO<sub>2</sub> and Zn doped CeO<sub>2</sub> thin films were deposited on the glass substrates (350°C) with two different Zn concentrations (4 and 8%). The prepared film has been characterized by X-ray Diffraction (XRD), Ultraviolet-Visible Spectroscopy (UV-Vis), and Scanning Electron Microscope (SEM). XRD analysis revealed that the films are well crystallized in nature having cubic fluorite structure with a grain orientation along (111) plane. CeO<sub>2</sub> thin films exhibits transparent in visible region and strong absorbance in the UV-Vis region. It was also observed that significant effect of Zn content on the transmittance and band gap energy of CeO<sub>2</sub> thin films. SEM micrograph shows that the particles are spherical in shape.

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INTRODUCTION

Cerium oxide (CeO<sub>2</sub>) thin films continuous attracting attention for scientists because of its cerium valence related optical property and its chemical stability used for many applications such as solid oxide fuel cells (SOFCs) (Steele, 1993; McAleese *et al.*, 1996), miniature capacitors (Nakazawa *et al.*, 1995; Roh *et al.*, 1997) and counter electrode in smart window devices. CeO<sub>2</sub> has unstable fluorite structure where the Ce<sup>4+</sup> has tendency to be transformed into Ce<sup>3+</sup>. These unique property combined with high UV absorbing ability, high refractive index and high transparency in the visible light region have been used in specific applications, including polishing material for glass, polymer and wood, UV-blocker and filter and as an additive to glass to protect sensitive materials (Goharshadi *et al.*, 2011; Karakoti *et al.*, 2008). Thin films can be prepared via many methods such as RF magnetron sputtering (Ryasnyanskiy *et al.*, 2007), pulsed layer deposition (PLD) (Reintjes, 1984), vacuum evaporation (Ahalapitiya, 2003) and spray pyrolysis (Oleynik and Adam, 2003). The high quality thin films prepared in this work are using a cost effective spray deposition technique.

However, the effects of zinc doped CeO<sub>2</sub> thin films have been rarely reported. In the present work, structural and optical properties of undoped and Zn doped CeO<sub>2</sub> thin films have been reported.

MATERIALS AND METHODS

The spray pyrolysis experimental arrangement has been described elsewhere (Addou *et al.*, 1999). The thin films of CeO<sub>2</sub> and Zn doped CeO<sub>2</sub> were deposited on to glass substrates, heated at 350° C from an aqueous solution containing hepta hydrate cerium chloride and zinc chloride. The spraying flow rate was set to 2ml/min and nozzle to substrate distance was about 20 cm. Before spraying, the glass substrates are properly cleaned and annealed at 100°C for 30 minutes, in order to eliminate water molecules. X-ray diffraction (XRD) patterns of samples were recorded with a “Xpert PRO” X-ray diffractometer using CuK $\alpha$  radiation. The optical transmittance and absorption characteristics were measured with a LAMDA 25 PERKIN ELMER spectrophotometer in the wavelength range of 300 – 800 nm. The surface morphologies of CeO<sub>2</sub> thin films were examined by SEM (JEOL Model, JSM-5610LV). The Mitutoyosurfstest SJ-301 is a stylus-type Instrument is used for measuring the thickness of the films.

RESULTS AND DISCUSSION

Structural characterization

Fig.1 shows the XRD patterns of undoped and Zn doped CeO<sub>2</sub> thin films consist of (111), (200), (220) and (311) diffraction peaks corresponding to  $2\theta = 28.6^\circ, 33^\circ, 46.9^\circ$  and  $56^\circ$  respectively. No other peaks belonging to other phases are detected by XRD. This indicates the phase purity of the films

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and also confirms that the doping of Zn does not affect the cubic fluorite structure  $\text{CeO}_2$ . It is clear that the peaks of the Zn doped  $\text{CeO}_2$  films were broadening and shifted toward higher angles as compared with pure  $\text{CeO}_2$  films. The broadening of the peaks indicates that  $\text{CeO}_2$  and Zn doped  $\text{CeO}_2$  films consist of nanometric size particle.

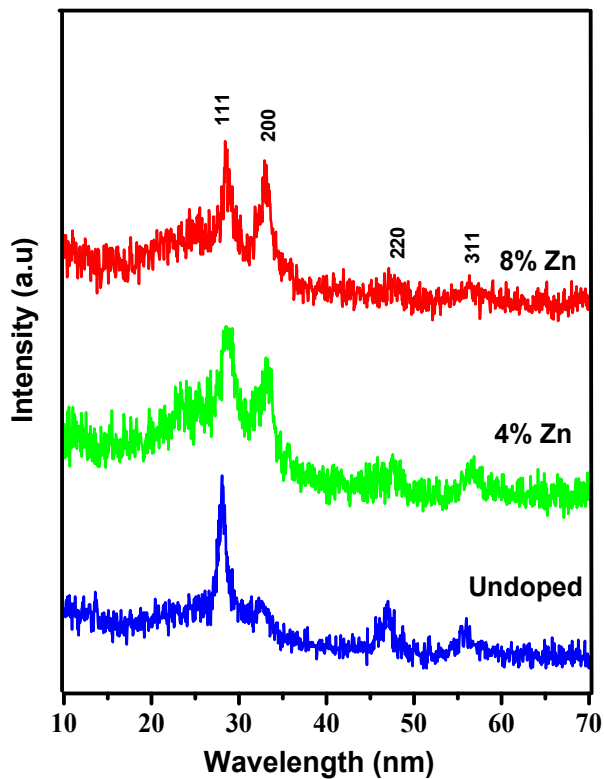


Fig. 1. XRD patterns for Undoped and Zinc doped  $\text{CeO}_2$  thin films prepared at  $350^\circ\text{C}$

Table 1. The Structural parameters of Undoped and Zn doped  $\text{CeO}_2$  thin films prepared at  $350^\circ\text{C}$

Mole concentration (M)	Thickness (t)	D (nm)	N ( $\times 10^{15}$ )	Dislocation density $\delta$ ( $\times 10^{14}$ lines/ $\text{m}^2$ )	Microstrain $\epsilon$ ( $\times 10^{-4}$ )	Lattice constant ( $\text{\AA}$ )	Volume of unit cell ( $\text{\AA}^3$ )
0.03	112	75	265	1.77	0.764	5.440	160.98
4% Zn	113	73	290	1.87	0.764	5.221	142.31
8% Zn	152	70	443	2.04	0.771	5.210	141.42

Table 2. Optical parameters for Undoped and Zn doped  $\text{CeO}_2$  thin films prepared at  $350^\circ\text{C}$

	Transmittance (at 600 nm)	Refractive Index (n) (at 600 nm)	Band gap (eV)
Undoped	78%	1.40	3.30
4% Zn	71%	1.72	3.20
8% Zn	65%	1.82	3.25

The calculated parameters from XRD data are given in Table 1. It was found that by Zn doping into  $\text{CeO}_2$  films, that the crystallite size and lattice parameters decreased. This is because the ionic radii of  $\text{Ce}^{3+}$  (0.103  $\text{\AA}$ ) and  $\text{Ce}^{4+}$  (0.092  $\text{\AA}$ ) are larger than that of  $\text{Zn}^{2+}$  (0.074  $\text{\AA}$ ), it opens up the possibilities of easy substitution of  $\text{Zn}^{2+}$  in the  $\text{CeO}$  lattice. The microstrain (Hankare *et al.*, 2009) increased with Zn, which may be due to the presence of defects and vacancies in the Zn – Ce – O lattice (Rong and Watkins, 1987).

## Optical properties

Fig.2 (a) and (b) shows the transmittance and absorbance spectra of undoped and Zn doped  $\text{CeO}_2$  thin films measured in the wavelength range of 300 to 800 nm. The optical transmittance of the film changed from 65% to 78% (at 600 nm) with the increase of Zn doping percentage. The higher transmittance observed in the films (8% Zn) is attributed to less light scattering of photons and structural homogeneity (Bougrine *et al.*, 2005). The extinction coefficient (k) can be obtained from the relationship (Elkorashy, 1989)

$$k = \frac{\alpha \lambda}{4\pi}$$

The variation of extinction coefficient with wavelength is shown in Fig.2 (d). Extinction coefficient is high in the wavelength range of 300–400 nm and low in the range of 400–800 nm. The rise and fall in the extinction coefficient is directly related to absorption of light (Prathap *et al.*, 2007). The refractive index n of the film is determined using the relationship (Giulio *et al.*, 1987).

$$N = \frac{1 + R^{1/2}}{1 - R^{1/2}}$$

Where R is the normal reflectance. Fig.2 (e) shows the variation of refractive index of  $\text{CeO}_2$  films with different wavelengths. It was observed that the refractive index increases in the UV region then decreases gradually from 2.27 to 1.69 in the high transmission range. The optical band gap of Zn doped  $\text{CeO}_2$  thin film is estimated from Tauc relationship (Bardeen *et al.*, 1956) and the optical parameters of the Zn doped  $\text{CeO}_2$  are presented in the Table 2.

$$(\alpha h\nu)^2 = A(h\nu - E_g)$$

Where  $E_g$  is the optical band gap and A is a constant. Fig.2 (f) shows that the graph between  $(\alpha h\nu)^2$  and the photon energy (h $\nu$ ) and indicates that the electronic transitions are direct in nature across the band gap of the films. The extrapolation of the linear part of the curves onto the energy axis allows to determine the optical band gap and its values varies in the range of 3.20 to 3.35 eV.

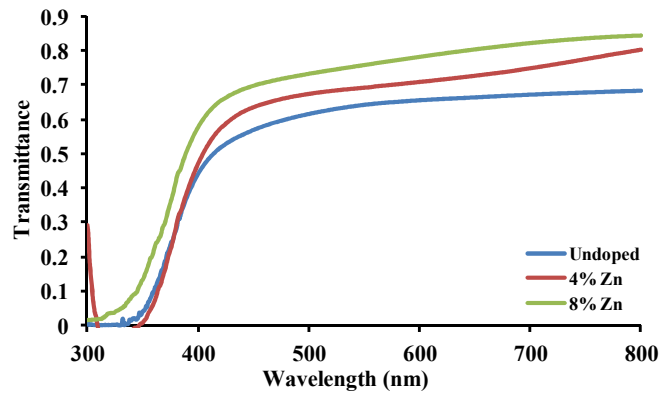


Fig. 2 (a). Transmittance vs Wavelength for Undoped and Zinc doped CeO<sub>2</sub> thin films prepared at 350°C

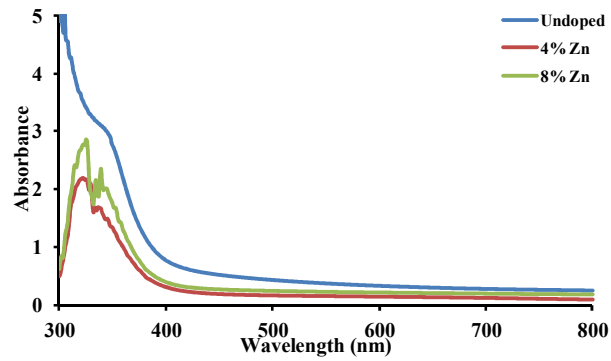


Fig. 2 (b). Absorbance vs Wavelength for Undoped and Zinc doped CeO<sub>2</sub> thin films prepared at 350°C

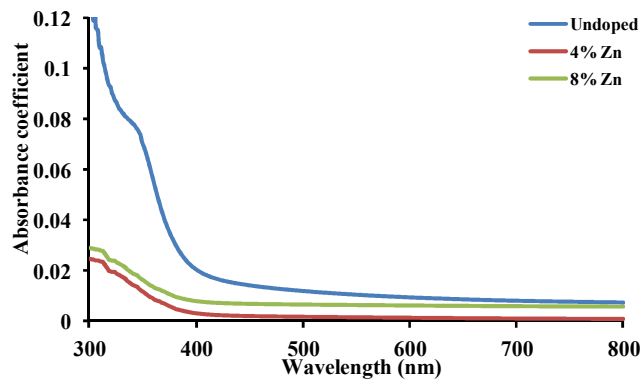


Fig. 2 (c). Absorbance coefficient vs Wavelength for Undoped and Zinc doped CeO<sub>2</sub> thin films prepared at 350°C

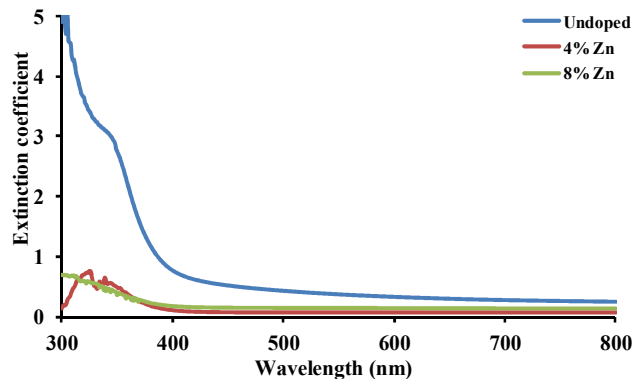


Fig. 2 (d). Extinction coefficient vs Wavelength for Undoped and Zinc doped CeO<sub>2</sub> thin films prepared at 350°C

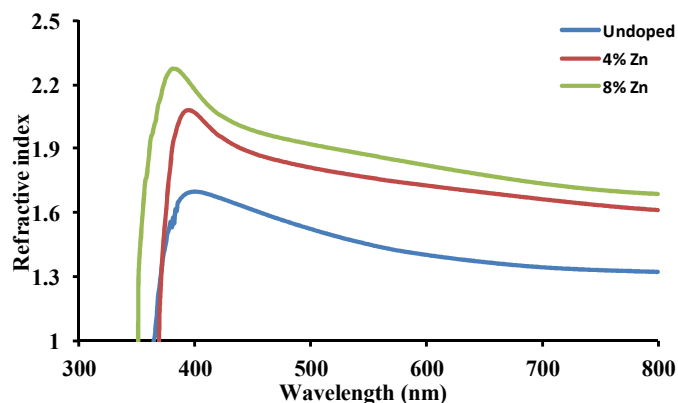


Fig.2 (e). Refractive index vs Wavelength for Undoped and Zinc doped  $\text{CeO}_2$  thin films prepared at  $350^\circ\text{C}$

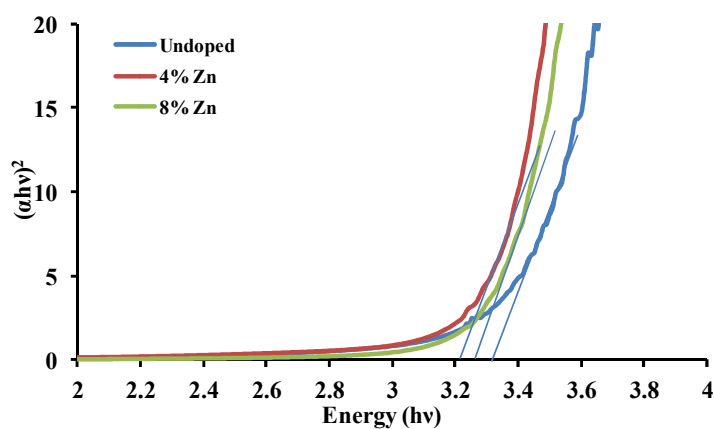


Fig. 2 (f). Energy (hv) vs  $(\alpha h\nu)^2$  for Undoped and Zinc doped  $\text{CeO}_2$  thin films prepared at  $350^\circ\text{C}$

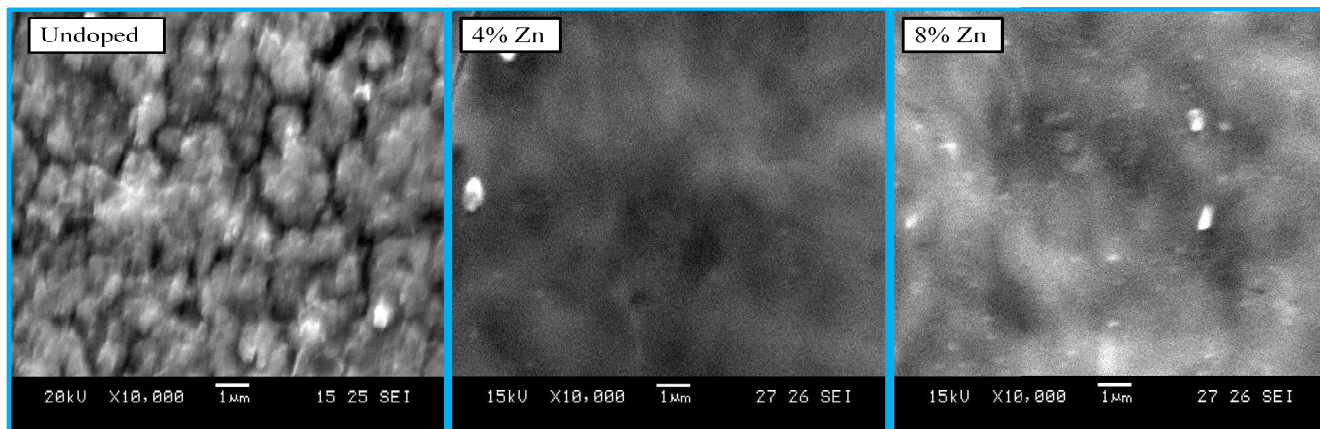


Fig. 3. SEM photographs for Undoped and Zinc doped  $\text{CeO}_2$  thin films prepared at  $350^\circ\text{C}$

### Morphological characterization

Surface morphology of undoped and Zn doped  $\text{CeO}_2$  thin films deposited at  $350^\circ\text{C}$  substrate temperature is shown in Fig.3. The undoped  $\text{CeO}_2$  film image shows big size aggregates of  $\text{CeO}_2$  particles without much individual particles. For Zn doped thin films, the image shows a smooth uniform surface oxide with fine grains.

### Conclusion

Undoped and Zn doped  $\text{CeO}_2$  thin films were successfully prepared by a simple and low cost spray pyrolysis technique. The structural analysis of the deposited film suggests that the thin films were cubic fluorite structure. UV-Visible spectra reveal that the transmittance of the films increased with increasing zinc concentration and the observed band gap are in the range of 3.20 to 3.35 eV. SEM image indicated that this

method is useful to prepare a high quality and microcrack free films. The deposited CeO<sub>2</sub> thin films are used for various applications such as solid oxide fuel cell, miniature capacitors and counter electrode in smart window devices.

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